

PLAUNOLIDE, A FURANOID DITERPENE FROM *CROTON SUBLYRATUS*

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Abstract—The isolation and structural elucidation of plaunolide, a new furanoid diterpene, from *Croton sublyratus* are described

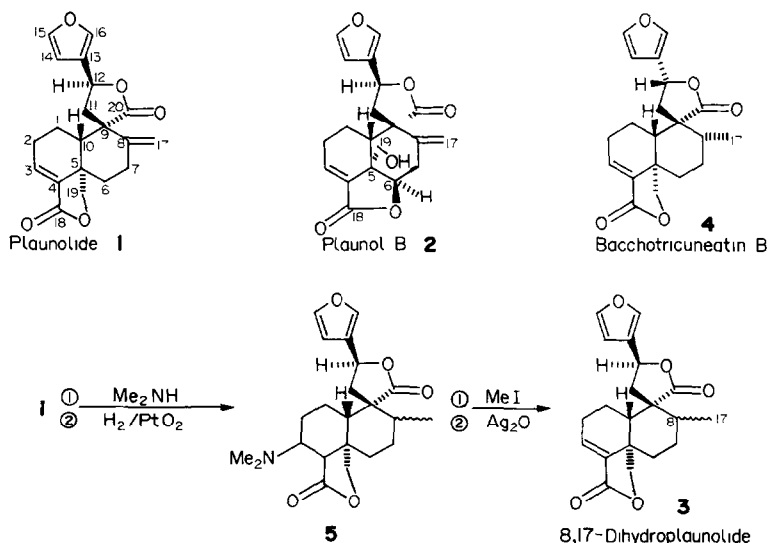
We have recently isolated and characterized diterpene alcohols [1–4] with antipeptic ulcer activity from a crude drug, named *plau-noi* in Thailand, which is prepared from stems of *Croton sublyratus* Kurz [5].

From the acetone extract of this plant we have isolated another major component named plaunolide (1). Compound 1, $C_{20}H_{20}O_5$ (mass spectral and elemental analysis), contained two γ -lactone ring systems (IR ν cm^{-1} 1775, 1750), a β -monosubstituted furan moiety [IR ν cm^{-1} 3140, 1505, 875, 1H NMR δ 7.64, 7.56, 6.50 (1H \times 3, narrow multiplet each), MS m/z 95, 94, 81] and an end methylene function [IR ν cm^{-1} 1645, 895, 1H NMR δ 5.06, 4.82 (1H \times 2, narrow doublet each)]. The spectral and chemical properties of 1 closely resembled those of plaunol B (2) [2, 3] which was isolated previously from the same plant and characterized as a furanoditerpene lactone of the ent-clerodane type. On comparison of the 1H NMR spectrum of 1 with that of 2, it was clear that the hydroxymethyl group at C-5 of 2 was attached to C-18 to

form a γ -lactone in the case of 1, as shown by the absence from 1 of a doublet of doublets due to H-6 (cf 2 δ 4.74, 1H, $J = 12, 7$ Hz) and a downfield shift of an AB-system attributable to H-19 [2 δ 3.90, 3.54 ($J = 11$ Hz), 1 δ 4.22, 4.15 ($J = 9$ Hz)]. All the data thus led to formula 1.

To compare 1 with bacchotricuneatin B isolated from *Baccharis tricuneata* (L.f.) Pers. var. *tricuneata* by H. Wagner *et al.* [6, 7], it was converted to 8,17-dihydroplaunolide (3) by the route 1 \rightarrow 5 \rightarrow 3. The 1H NMR spectrum of 3, as well as the other spectra, closely resembled that of bacchotricuneatin B (4) except for a chemical shift due to Me-17 [3 δ 1.05 ($d, J = 6$ Hz), 4 δ 1.14 ($d, J = 6$ Hz)]. This indicated that 8,17-dihydroplaunolide was a stereoisomer of 4. To confirm this result, X-ray analysis of plaunolide was undertaken [8]. The structure of plaunolide, solved by the program MULTAN (the final R-factor 0.069), was found to correspond to formula 1.

Finally, the absolute configuration of plaunolide was



determined to be as shown in formula 1 by comparison of the negative Cotton effect in the CD ($\pi \rightarrow \pi^*$ band) of 3 ($[\theta]_{245}^{\text{MeCN}} - 75\,200$) with that of 4 ($[\theta]_{245}^{\text{MeCN}} - 17\,600$) [7]

EXPERIMENTAL

All mps were uncorr ^1H NMR TMS as an int reference

Extraction and isolation Crushed crude drug (81.5 kg) was extracted ($\times 3$) with Me_2CO under reflux. After evaporation of the solvent, the residue was fractionated as in ref [3] to give plaunolide (220 g) after Si gel chromatography [C_6H_6 -EtOAc (1:1), R_f 0.37] Mp 169–172°, $[\alpha]_D^{25} - 75.2^\circ$ (Me_2CO , c 1.04), MS (75 eV) m/z (rel int) 340 [$\text{M}]^+$ (76), 310 (100), 292 (20), 229 (43), 149 (42), 95 (12), 94 (6), 81 (3), IR $\nu_{\text{max}}^{\text{nujol}} \text{cm}^{-1}$ 3140, 1775, 1755, 1670, 1645, 1605, 1505, 895, 875, ^1H NMR (100 MHz, CDCl_3) δ 7.64 (1H, *m*), 7.56 (1H, *m*), 6.64 (1H, *dd*, $J = 7.4$ Hz), 6.50 (1H, *m*), 5.69 (1H, *dd*, $J = 7.8$ Hz), 5.06 (1H, *d*, $J = 2$ Hz), 4.82 (1H, *br s*), 4.22, 4.15 (AB system $J = 9$ Hz), 2.02 (1H, *d*, $J = 8$ Hz), 1.58 (1H, *m*) [Found C, 70.10; H, 5.90 $\text{C}_{20}\text{H}_{20}\text{O}_5$ requires C, 70.58, H, 5.92%]

8,17-Dihydroplaunolide (3) A soln of 2.0 g plaunolide (1) and 5 ml dimethylamine in 40 ml THF was kept at room temp overnight. After addition of H_2O and EtOAc, the EtOAc layer was extracted with 5% HCl. The aq. HCl layer was made basic with NaHCO_3 and extracted with EtOAc. After evaporation of the solvent, the residue was recrystallized from CH_2Cl_2 -Et $_2\text{O}$ to give 1.8 g of addition product, mp 205–207°, of which 1.35 g was dissolved in 50 ml EtOAc and hydrogenated over 50 mg PtO_2 . The usual work-up gave 1.3 g 5, mp 210–212°. A mixture of 2 g 5 and 5 ml MeI in 20 ml MeCN was heated to reflux for 2 hr. Usual work-up gave 2.2 g of the desired product, mp 228–230°. To a soln of 1.0 g of this product in 10 ml THF and 10 ml H_2O was added Ag_2O (freshly prepared from 2 g AgNO_3 and 0.5 g NaOH), and the mixture was heated to 60° for 3 hr. Usual work-up gave 0.7 g 3. Mp 189–190°, $[\alpha]_D^{25} - 108.6^\circ$ (CHCl_3 , c 0.625), MS (75 eV) m/z (rel int) 342 [$\text{M}]^+$ (0.7), 312 (100), 284 (7.4), 267

(5.1), 239 (6.2), 218 (8.0), 190 (10.5), 173 (8.6), 145 (13.3), 105 (12.6), 95 (26.5), 94 (25.9), 91 (23.7), 81 (14.4), IR $\nu_{\text{max}}^{\text{KBr}} \text{cm}^{-1}$ 1755, 1660, 1600, 1510, 1450, 1320, 1195, 1170, 1135, 975, 925, 875, ^1H NMR (100 MHz, CDCl_3) δ 7.46 (1H, *m*), 7.44 (1H, *m*), 6.70 (1H, *dd*, $J = 6.3$ Hz), 6.40 (1H, *m*), 5.38 (1H, *t*, $J = 8.5$ Hz), 4.68, 3.85 (2H, ABX, $J = 9.5, 2$ Hz), 2.6–1.2 (12H, *m*), 1.05 (3H, *d*, $J = 6$ Hz) (Found C, 69.87, H, 6.59 $\text{C}_{20}\text{H}_{22}\text{O}_5$ requires C, 70.16, H, 6.48%)

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